

Time-Domain Modeling of Excited State Dynamics in Halide Perovskites

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Photo-induced processes play key roles in photovoltaic and photo-catalytic applications of halide perovskites, requiring understanding of the material's dynamical response to the photo-excitation on atomic and nanometer scales. Our non-adiabatic molecular dynamics techniques,¹ implemented within time-dependent density functional theory,²⁻⁴ allow us to model such non-equilibrium response in the time domain and at the atomistic level. Considering realistic aspects of perovskite structure,⁵ we demonstrate that strong interaction at the perovskite/TiO₂ interface facilitates ultrafast charge separation,⁶ how dopants can be used to both decrease and increase charge recombination,⁷⁻⁹ that grain boundaries constitute a major reason for charge losses,⁹ that moderate humidity increases charge lifetime, while high humidity accelerates losses,¹⁰ that hole trapping by iodine interstitial, surprisingly, extends carrier lifetime,¹¹ that collective nature of dipole motions inhibits nonradiative relaxation,¹² that organic cation orientation has a strong effect on inorganic ion diffusion and current-voltage hysteresis,¹³ that surface passivation with Lewis base molecules decelerates nonradiative charge recombination by an order of magnitude,¹⁴ that the experimentally observed dual (hot/cold) fluorescence originates from two types of perovskites substructures,¹⁵ that doping with larger cations increases lattice stiffness and slows down nonradiative charge recombination,¹⁶ why PbI₂ rich perovskites show better performance,¹⁷ that halide composition can be used to control charge relaxation,¹⁸ that oxidation states of halide defects strongly influence charge trapping and recombination,¹⁹ and why perovskites exhibit unusual temperature dependence of electron and hole lifetimes.²⁰

1. L. J. Wang, A. Akimov and O. V. Prezhdo, *J. Phys. Chem. Lett.*, 2016, **7**, 2100-2112.
2. A. V. Akimov and O. V. Prezhdo, *J. Chem. Theor. Computation*, 2013, **9**, 4959-4972.
3. A. V. Akimov and O. V. Prezhdo, *J. Chem. Theory Comput.*, 2014, **10**, 789-804.
4. S. Pal, D. J. Trivedi, A. V. Akimov, B. Aradi, T. Frauenheim and O. V. Prezhdo, *J. Chem. Theory Comput.*, 2016, **12**, 1436-1448.
5. J. Jankowska, R. Long and O. V. Prezhdo, *ACS Energy Lett.*, 2017, **2**, 1588-1597.
6. R. Long, W. H. Fang and O. V. Prezhdo, *J. Phys. Chem. C*, 2017, **121**, 3797-3806.
7. J. Liu and O. V. Prezhdo, *J. Phys. Chem. Lett.*, 2015, **6**, 4463-4469.
8. R. Long and O. V. Prezhdo, *ACS Nano*, 2015, **9**, 11143-11155.
9. R. Long, W. H. Fang and O. V. Prezhdo, *J. Phys. Chem. Lett.*, 2016, **7**, 3215-3222.
10. R. Long, J. Liu and O. V. Prezhdo, *J. Am. Chem. Soc.*, 2016, **138**, 3884-3890.
11. W. Li, J. Liu, F. Q. Bai, H. X. Zhang and O. V. Prezhdo, *ACS Ener Lett.*, 2017, **2**, 1270.
12. J. Jankowska and O. V. Prezhdo, *J. Phys. Chem. Lett.*, 2017, **8**, 812-818.
13. C. J. Tong, W. Geng, O. V. Prezhdo and L. M. Liu, *ACS Energy Lett.*, 2017, **2**, 1997.
14. L. H. Liu, W. H. Fang, R. Long and O. V. Prezhdo, *J. Phys. Chem. Lett.*, 2018, **9**, 1164.
15. Z. S. Zhang, R. Long, M. V. Tokina and O. V. Prezhdo, *J. Am. Chem. Soc.*, 2017, **139**, 17327-17333.
16. J. L. He, W. H. Fang, R. Long and O. V. Prezhdo, *ACS Energy Lett.*, 2018, **3**, 2070.
17. C. J. Tong, L. Q. Li, L. M. Liu and O. V. Prezhdo, *ACS Energy Lett.*, 2018, **3**, 1868.
18. J. L. He, A. S. Vasenko, R. Long and O. V. Prezhdo, *J. Phys. Chem. Lett.*, 2018, **9**, 1872.
19. W. Li, Y. Y. Sun, L. Q. Li, Z. H. Zhou, J. F. Tang and O. V. Prezhdo, *J. Am. Chem. Soc.*, 2018, **140**, 15753.
20. W. Li, J. F. Tang, D. Casanova and O. V. Prezhdo, *ACS Energy Lett.*, 2018, **3**, 2713.